

### Naming Binary Ionic Compounds

Formula of the Ionic Compound	Name of the Ionic Compound
AlP	Aluminum phosphide
MgS	Magnesium sulfide
CaO	Calcium oxide
AlN	Aluminum nitride
BaS	Barium sulfide
KCl	Potassium chloride
LiF	Lithium fluoride
Na <sub>2</sub> S	Sodium sulfide
BaH <sub>2</sub>	Barium hydride
<del>Al<sub>3</sub>N<sub>2</sub></del> <b>AlN</b>	Aluminum nitride
SrF <sub>2</sub>	Strontium fluoride
Cs <sub>2</sub> S	Cesium sulfide
MgCl <sub>2</sub>	Magnesium chloride
K <sub>3</sub> P	Potassium phosphide
BaI <sub>2</sub>	Barium iodide
Li <sub>3</sub> P	Lithium phosphide
BaI <sub>2</sub>	Barium iodide
AlCl <sub>3</sub>	Aluminum chloride
KH	Potassium hydride
Ca <sub>3</sub> P <sub>2</sub>	Calcium phosphide
CaS	Calcium sulfide
Li <sub>3</sub> N	Lithium nitride
SrS	Strontium sulfide