

NOMENCLATURE CHEMISTRY 1406

**BINARY IONIC**

KOH \_\_\_\_\_

Li<sub>2</sub>O \_\_\_\_\_

KCl \_\_\_\_\_

NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> \_\_\_\_\_

LiH \_\_\_\_\_

NaHCO<sub>3</sub> \_\_\_\_\_

Ba(BrO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

Al(OH)<sub>3</sub> \_\_\_\_\_

NH<sub>4</sub>NO<sub>3</sub> \_\_\_\_\_

NaClO<sub>4</sub> \_\_\_\_\_

NaClO<sub>2</sub> \_\_\_\_\_

Be(IO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

CaHCO<sub>3</sub> \_\_\_\_\_

BeO \_\_\_\_\_

Na<sub>2</sub>C<sub>2</sub>O<sub>4</sub> \_\_\_\_\_

KH \_\_\_\_\_

Al(OH)<sub>3</sub> \_\_\_\_\_

NaSCN \_\_\_\_\_

KClO<sub>4</sub> \_\_\_\_\_

Ca(NO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

CaCl<sub>2</sub> \_\_\_\_\_

Na<sub>2</sub>SO<sub>3</sub> \_\_\_\_\_

CaBr<sub>2</sub> \_\_\_\_\_

CaC<sub>2</sub> \_\_\_\_\_

NH<sub>4</sub>OH \_\_\_\_\_

Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> \_\_\_\_\_

CsOH \_\_\_\_\_

MgSO<sub>4</sub> \_\_\_\_\_

Na<sub>2</sub>SO<sub>4</sub> \_\_\_\_\_

Li<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_

AlBr<sub>3</sub> \_\_\_\_\_

Sr(NO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

BaS \_\_\_\_\_

KSCN \_\_\_\_\_

NH<sub>4</sub>Cl \_\_\_\_\_

K<sub>2</sub>S<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

KCl \_\_\_\_\_

SrS \_\_\_\_\_

Al<sub>2</sub>(CrO<sub>4</sub>)<sub>3</sub> \_\_\_\_\_

Mg(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> \_\_\_\_\_

NaF \_\_\_\_\_

(NH<sub>4</sub>)<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> \_\_\_\_\_